# Structure and Mechanism of the Influenza A M2 18-60 Dimer of Dimers 

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(s) Supporting Information


#### Abstract

We report a magic angle spinning (MAS) NMR structure of the drug-resistant S31N mutation of M2 ${ }_{18-60}$ from Influenza $A$. The protein was dispersed in diphytanoyl-sn-glycero-3-phosphocholine lipid bilayers, and the spectra and an extensive set of constraints indicate that $\mathrm{M}_{18-60}$ consists of a dimer of dimers. In particular, $\sim 280$ structural constraints were obtained using dipole recoupling experiments that yielded well-resolved ${ }^{13} \mathrm{C}-{ }^{15} \mathrm{~N},{ }^{13} \mathrm{C}-{ }^{13} \mathrm{C}$, and ${ }^{1} \mathrm{H}-{ }^{15} \mathrm{~N} 2 \mathrm{D}, 3 \mathrm{D}$, and 4D MAS spectra, all of which show cross-peak doubling. Interhelical distances were measured using mixed ${ }^{15} \mathrm{~N} /{ }^{13} \mathrm{C}$ labeling and with deuterated protein, MAS at $\omega_{\mathrm{r}} / 2 \pi=60 \mathrm{kHz}$,  $\omega_{0 \mathrm{H}} / 2 \pi=1000 \mathrm{MHz}$, and ${ }^{1} \mathrm{H}$ detection of methyl-methyl contacts. The experiments reveal a compact structure consisting of a tetramer composed of four transmembrane helices, in which two opposing helices are displaced and rotated in the direction of the membrane normal relative to a four-fold symmetric arrangement, yielding a two-fold symmetric structure. Side chain conformations of the important gating and pH -sensing residues W41 and H37 are found to differ markedly from four-fold symmetry. The rmsd of the structure is $0.7 \AA$ for backbone heavy atoms and $1.1 \AA$ for all heavy atoms. This two-fold symmetric structure is different from all of the previous structures of M2, many of which were determined in detergent and/or with shorter constructs that are not fully active. The structure has implications for the mechanism of $\mathrm{H}^{+}$transport since the distance between His and Trp residues on different helices is found to be short. The structure also exhibits two-fold symmetry in the vicinity of the binding site of adamantyl inhibitors, and steric constraints may explain the mechanism of the drug-resistant S31N mutation.


## INTRODUCTION

Influenza A M2 is a 97 residue transmembrane protein that assembles as a tetramer ${ }^{1}$ and conducts protons at low $\mathrm{pH}^{2}$ in order to trigger membrane fusion in an endosome and unpacking of the viral genome. ${ }^{3,4}$ The N-terminus of the protein is positioned on the outside of infected cells, with at least the first 18 residues exposed. ${ }^{5,6}$ A single-pass $\alpha$-helix places the C-terminal domain on the cytoplasmic side of infected cells, a portion of which forms an amphipathic helix responsible for stabilization of the protein. ${ }^{7}$

The $\mathrm{M} 2 \mathrm{H}^{+}$transporter is critical for viral replication, as evidenced by the therapeutic effect of aminoadamantyl inhibitors known to target M2 and to reduce proton conduction. ${ }^{7}$ There are now several excellent structures of the wild-type (WT) protein in complex with aminoadamantyl inhibitors in the pore, placing the site of pharmacological
binding between residues 27 and $34 .{ }^{8-12}$ However, resistance has developed in circulating strains of influenza, primarily due to a single-point mutation S31N, which has precipitated a need for new inhibitors that target S31N M2. Since the proton conduction process in M2 is very sensitive to the structure of the peptide, this motivates the determination here of the structure of S31N M2 in hydrated lipid bilayers.

Previous structural studies of M2 have been applied to several different constructs reconstituted in a variety of virus membrane mimetics. These constructs can be classified into three groups, the full-length protein (FL), the conduction domain (CD), and the transmembrane domain (TM). The CD comprises approximately residues $18-60$, which includes both

[^0]the transmembrane residues $(25-46)$ as well as an amphipathic helix ( $\sim 47-59$ ), which is known to stabilize the tetrameric assembly. ${ }^{7,13}$ TM domain constructs from residue 22 to 46 contain a single membrane-spanning helix that does not fully reproduce the conduction and drug inhibition of WT but remains drug-sensitive. Since the CD forms tetramers with conduction and drug sensitivity that are indistinguishable from that of the full protein, we have investigated a CD construct, $\mathrm{M} 2_{18-60}{ }^{13}$

Within the core of the tetrameric bundle, the four His-H37 residues are known to control the pH -dependent rate of conduction. Measurements on $\mathrm{M}_{22}-46^{14,15}$ and $\mathrm{M} 2_{18-60}{ }^{16}$ showed a multiplicity of $\mathrm{H} 37 \mathrm{p} K_{\mathrm{a}}$ values and pinpointed the third protonation event as being responsible for physiological conduction. Thus, although the channel is a homotetramer, H37 exists as a mix of imidazole and imidazolium, ${ }^{17}$ implying that the highest degree of symmetry for doubly protonated M2 tetramers is two-fold. In lipid bilayers, we find that two-fold symmetry is detected even for $\mathrm{M} 2_{18-60}$ tetramers that are uncharged at H37. ${ }^{18-20}$

Magic angle spinning (MAS), oriented sample (OS), and solution NMR have been used extensively to study various constructs of M2. The initial solid-state experiments were published by Cross and co-workers, ${ }^{21-24}$ and more recently, Hong and co-workers ${ }^{11,12,14,25}$ have made important contributions to the study of constructs of M2. Extensive solution NMR studies have been described by the groups of Chou ${ }^{9,13,26}$ and DeGrado. ${ }^{27-30}$

Here we report MAS NMR studies of the M2 $1_{18-60}$ construct carrying the drug-resistant mutation S31N from Influenza A dispersed in diphytanoyl-sn-glycero-3-phosphocholine lipid bilayers. We employed a number of dipole recoupling experiments that yielded well-resolved ${ }^{13} \mathrm{C}-{ }^{15} \mathrm{~N}$ and ${ }^{13} \mathrm{C}-{ }^{13} \mathrm{C}$ 2D and 3D MAS spectra that permit determination of structural constraints for membrane protein samples. Samples were isotopically labeled with several different schemes, including uniform ${ }^{13} \mathrm{C}$ and ${ }^{15} \mathrm{~N}$ labeling, labeling by residue type, and site-specific ${ }^{13} \mathrm{C}$ labeling using $1,6-{ }^{13} \mathrm{C}$ glucose. Interhelical distances were measured unambiguously using mixed ${ }^{15} \mathrm{~N} /{ }^{13} \mathrm{C}$ labeling. Several additional restraints were determined using extensively deuterated protein, MAS at $\omega_{\mathrm{r}} / 2 \pi$ $=60 \mathrm{kHz}$ and $\omega_{0 \mathrm{H}} / 2 \pi=1000 \mathrm{MHz}$, and ${ }^{1} \mathrm{H}$ detection of methyl-methyl contacts in 3D and 4D MAS experiments. A mechanistically important ${ }^{15} \mathrm{~N}-{ }^{1} \mathrm{H}-{ }^{15} \mathrm{~N}$ distance of $<3.5 \AA$ was observed between His and Trp on the two different chains of the dimer. The experimental protocol could represent a paradigm for structural determinations of membrane protein samples. Collectively, these experiments yielded 283 structural constraints and permitted us to calculate a structure consisting of four compact transmembrane helices. However, importantly, two opposing helices are displaced and rotated in the direction of the membrane normal, yielding a two-fold symmetric structure. Side chain conformations of the important gating and pH -sensing residues W 41 and H 37 are found to differ markedly from four-fold symmetry. The rmsd of the structure is $0.7 \AA$ for backbone and $1.1 \AA$ for all heavy atoms. This two-fold symmetric structure is clearly different from all of the previously reported structures of WT M2 and from the two previously reported structures of S31N M2, ${ }^{26,29}$ which were solved in detergent micelles.

The dimer of dimers structure reported here was determined in lipid bilayers that closely mimic the viral environment. We used a preparation that allowed pharmacological drug binding
in WT M2, ${ }^{8}$ supporting the assertion that we have reconstituted a native state of the protein. The structure calculated from the MAS NMR constraints therefore has implications for the path of $\mathrm{H}^{+}$transportation and inhibitor binding. In particular, there is direct contact between a His and $\operatorname{Trp}$ on different helices, which is likely important in $\mathrm{H}^{+}$ transport. Furthermore, constriction within the tetramer at residues H37 and W41 excludes passage of water, consistent with the current understanding that M2 conducts via $\mathrm{H}^{+}$ shuttling. Finally, the structure also exhibits two-fold symmetry in the vicinity of the site of adamantyl inhibitors, and comparison of steric constraints with drug-bound structures may explain the mechanism of the drug-resistant S31N mutation.

## EXPERIMENTAL SECTION

M2 Synthesis. Recombinant M2 was synthesized, purified, and refolded in 1,2-diphytanoyl-sn-glycero-3-phosphocholine (DPhPC) bilayers at a lipid/protein ratio of $1: 1$ by weight as described previously. ${ }^{18}$ Briefly, overexpression was carried out in Escherichia coli BL21 DE3 $3^{13,18}$ in minimal media prepared per liter with $3-4 \mathrm{~g}$ of glucose, 1 g of ammonium chloride, standard salts, and Centrum adult vitamins ( 1.5 mL of 2 pills in 40 mL , dissolved with shaking for 30 min ). The construct used was an N -terminal fusion to TrpLE, with a His- 9 tag. After nickel affinity purification under denaturing conditions of 6 M guanidine, the fusion was precipitated by dialysis against water. The fusion was cleaved in $70 \%$ formic acid using cyanogen bromide ( $1-3 \mathrm{~h}$ at $\sim 20^{\circ} \mathrm{C}$ ), diluted, and lyophilized. The resulting powder was dissolved in 2:1:1 formic acid/HFIP/water and injected onto a C4 column equilibrated at $5 \%$ isopropyl alcohol. The protein eluted near the end of a gradient from $5 \%$ isopropyl alcohol to 58:37:5 acetonitrile/2-propanol/water. All reverse-phase chromatography solutions had $0.1 \%$ trifluoroacetic acid. The pure protein was lyophilized. Lyophilized protein was refolded by dissolving the protein in octyl glucoside detergent solution, mixing with a lipid solution dissolved in minimal octyl glucoside, and removal of detergent by dialysis at $\sim 20^{\circ} \mathrm{C}$ against sample buffer ( 40 mM phosphate, 30 mM glutamate, 3 mM sodium azide, pH 7.8 ). For samples with mixed labeling, cells containing the differently labeled protein were mixed and purified together for convenience.

Deuterated and ILV methyl-labeled protein was produced as described below and reconstituted in ${ }^{2} \mathrm{H}_{78}$ DPhPC. ${ }^{2} \mathrm{H}_{78}$ DPhPC was synthesized in collaboration with FBReagents (Cambridge, MA). Details are provided in the Supporting Information.
$U-{ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$ M2. Per liter, 3 g of ${ }^{13} \mathrm{C}$ glucose and 1 g of ${ }^{15} \mathrm{~N}$ ammonium chloride (CIL), with another $0.5-1 \mathrm{~g}$ of ${ }^{13} \mathrm{C}$ glucose added at induction ( $150 \mu \mathrm{M}$ IPTG, $18{ }^{\circ} \mathrm{C}$ ), were added at an $\mathrm{OD}_{600}$ of $0.7-0.9$. The final $\mathrm{OD}_{600}$ reached 3.5-4 after 18 h of expression at 18 ${ }^{\circ} \mathrm{C}$ and yielded up to 15 mg of pure protein.
$U-{ }^{13} C_{1}^{15} N-\left[^{12} C,{ }^{74} N-\right.$ ILFY] M2. Per liter, 3 g of ${ }^{13} \mathrm{C}$ glucose and 1 g of ${ }^{15} \mathrm{~N}$ ammonium chloride (CIL), with another $0.5-1 \mathrm{~g}$ of ${ }^{13} \mathrm{C}$ glucose, were added at induction. Cells were grown in 700 mL of media, with 300 mL of media used to dissolve the amino acids. When the $\mathrm{OD}_{600}$ reached 0.7 , the reserved 300 mL of media was added, along with amino acids at natural isotopic abundance: ${ }^{31,32} 150 \mathrm{mg}$ of isoleucine, 150 mg of leucine, 50 mg of phenylalanine, and 100 mg of tyrosine (Sigma-Aldrich). Expression of protein was induced with $150 \mu \mathrm{M}$ IPTG at $18{ }^{\circ} \mathrm{C}$ as before. The final $\mathrm{OD}_{600}$ reached $3.5-4.6$ after 18 h of expression at $18{ }^{\circ} \mathrm{C}$ and yielded up to 15 mg of pure protein.
$\left[1-{ }_{-13} \mathrm{C}\right]$ Glucose-, $\left[1,6-{ }^{13} \mathrm{C}_{2}\right]$ Glucose M2. Preparation was identical to that for $\mathrm{U}-{ }^{13} \mathrm{C}-{ }^{-15} \mathrm{~N}$ M2 except that specifically isotopically labeled glucose was used. The labeling pattern resulting from $\left[1,6-^{13} \mathrm{C}\right]$ glucose is the same as that from $\left[1-{ }^{13} \mathrm{C}\right]$ glucose ${ }^{33}$ (Omicron Biochemicals) except that the $\left[1,6-{ }^{13} \mathrm{C}\right]$ glucose results in double the labeling level per site.
U- $-{ }^{13} C_{1}^{15} N_{1}^{2} \mathrm{H}-{ }^{12} \mathrm{C}_{1}^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H} \delta 1$-Ile, ${ }^{12} \mathrm{C},{ }_{1}^{13} \mathrm{Ca},{ }^{13} \mathrm{C}^{\prime},{ }^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H} \delta 2$-Leu, ${ }^{12} \mathrm{C}_{1}{ }^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H} \gamma 2$-Val] M2. An overnight culture in 50 mL of $90 \%{ }^{2} \mathrm{H}$ M9, $10 \%$ SOC media, was grown from a fresh transformation in BL21-

DE3 until reaching an $\mathrm{OD}_{600}$ of 2-3. Cells were then pelleted by centrifugation and transferred to 1 L of ${ }^{2} \mathrm{H}$ M9 media containing 3 g of ${ }^{2} \mathrm{H}-{ }^{13} \mathrm{C}$ glucose, 1 g of ${ }^{15} \mathrm{~N}$ ammonium chloride, salts, and Centrum as before, in 1 L of $99.8 \%{ }^{2} \mathrm{H} \mathrm{D}_{2} \mathrm{O}$. The doubling time was 2 h . At an $\mathrm{OD}_{600}$ of $0.65-0.75$, ILV precursors were added: 75 mg of $\alpha$ ketobutyric acid, ${ }^{34}$ sodium salt $4-{ }^{13} \mathrm{C} 99 \%, 3,3,4,4-{ }^{2} \mathrm{H}_{4}, 98 \%$ (CIL), and 350 mg of $2-\left({ }^{13} \mathrm{C},{ }^{2} \mathrm{H}_{2}\right)$ methyl-4- $\left({ }^{2} \mathrm{H}_{3}\right)$-acetolactate prepared as described previously. ${ }^{35}$ Cells were harvested after 24-36 h and yielded up to 5 mg of pure protein. This labeling pattern results in methyl groups with isolated ${ }^{13} \mathrm{C}^{1} \mathrm{H}$ spin pairs in an otherwise ${ }^{2} \mathrm{H},{ }^{12} \mathrm{C}$ background at nearby sites. The protein was purified and refolded in ${ }^{1} \mathrm{H}$ buffers, resulting in complete exchange to ${ }^{1} \mathrm{H}$ at exchangeable sites such as the backbone amides.
$U-{ }^{13} \mathrm{C}_{1}^{15} \mathrm{~N}_{,}^{2} \mathrm{H}-\left[{ }^{13} \mathrm{CH}_{3} \delta 1\right.$-Ile] M2. The same protocol as above was used, but the precursor was 75 mg of ${ }^{13} \mathrm{C} 4,98 \%, 3,3-{ }^{2} \mathrm{H}_{2}, 98 \% \alpha$ ketobytryic acid, and sodium salt.
$U-{ }^{13} \mathrm{C}_{1}^{15} \mathrm{~N}_{1}^{2} \mathrm{H}-\left[{ }^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H} \delta 2\right.$-Leu, ${ }^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H} \gamma 2$-Val] M2. The same protocol as above was used, but the precursor was 350 mg of $\mathrm{U}-{ }^{13} \mathrm{C}, 2-\left({ }^{2} \mathrm{H}_{2}\right)$ methyl-4- $\left({ }^{2} \mathrm{H}_{3}\right)$-acetolactate.

NMR Spectroscopy. NMR spectra were recorded on several highfield Bruker (Bruker Biospin, Billerica, MA) spectrometers operating at a ${ }^{1} \mathrm{H}$ frequency of 800,900 , and 1000 MHz , and using 3.2 mm Efree probes at 800 and 900 MHz and 1.3 mm probes at 800 and 1000 MHz . Spectra were also recorded using a home-built spectrometer designed by Dave Ruben and operating at a ${ }^{1} \mathrm{H}$ frequency of 750 MHz equipped with a Bruker 3.2 mm E-free probe. ${ }^{13} \mathrm{C}$-detected spectra were recorded with 3.2 mm Bruker E-free probes, and ${ }^{1} \mathrm{H}$-detected spectra were recorded with Bruker 1.3 mm solenoid probes tuned to HCN or HCD. The sample temperature was maintained at $20-30^{\circ} \mathrm{C}$ using Bruker cooling units (BCU II or BCU extreme) or on the 750 MHz instrument, a Kinetics Thermal System XR air-jet system (Stone Ridge, NY). Chemical shifts are reported on the DSS scale using adamantane as a secondary reference.

Spectral Assignments. The first task in any NMR structural determination involves assigning the spectra. In the case of the dimer of dimers of $\mathrm{M} 2_{18-60}$, sequence-specific chemical shift assignments of ${ }^{13} \mathrm{C}$ and ${ }^{15} \mathrm{~N}$ were determined from $3 \mathrm{D}{ }^{15} \mathrm{~N}-{ }^{13} \mathrm{C}-{ }^{13} \mathrm{C}$ correlations and $2 \mathrm{D}{ }^{13} \mathrm{C}-{ }^{13} \mathrm{C}$ correlations. Assignment of amide ${ }^{1} \mathrm{H}$ and backbone ${ }^{13} \mathrm{C}$ and ${ }^{15} \mathrm{~N}$ resonances was made independently using a suite of $\operatorname{six}{ }^{1} \mathrm{H}$ detected spectra and automated analysis via MATCH, ${ }^{36}$ as recently reported. ${ }^{37}$ These backbone resonance assignments were in agreement with those made manually using ${ }^{13} \mathrm{C}$ detection. Stereospecific methyl protons were assigned using ${ }^{13} \mathrm{C}-{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}-{ }^{13} \mathrm{C}-{ }^{1} \mathrm{H}$ correlations and the known ${ }^{13} \mathrm{C}$ shifts. Since many of the approaches we employed are new to MAS experiments, the spectra we used for assignment are detailed below.

NCACX and NCOCX connectivity was determined using 2D ${ }^{15} \mathrm{~N}-{ }^{13} \mathrm{C}$ ZF-TEDOR ${ }^{38}$ and $3 \mathrm{D}{ }^{15} \mathrm{~N}-{ }^{13} \mathrm{C}-{ }^{13} \mathrm{C}$ TEDOR-RFDR ${ }^{18,39-41}$ spectra recorded on $\mathrm{U}-{ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$ M2 and $\mathrm{U}-{ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}-\left[{ }^{12} \mathrm{C},{ }^{14} \mathrm{~N}\right.$-ILFY $]$ M2 as described previously. ${ }^{18}$ Spectra were recorded at a field of 900 MHz $\left({ }^{1} \mathrm{H}\right)$ and 20 kHz spinning, using 83 kHz RFDR pulses, $\sim 100 \mathrm{kHz}{ }^{1} \mathrm{H}$ decoupling during RFDR mixing, and 83 kHz decoupling during acquisition. Mixing periods for ZF-TEDOR and RFDR of 1.2 and 4.8 ms were used with XY-4 and XY-16 or XY-32 phase cycling for TEDOR and RFDR, respectively. The $\mathrm{N}-{ }^{13} \mathrm{C} \alpha$ plane from the TEDOR is shown in Figure 1, and selected strips from the 3D are shown in Figure 2. In Figure 1, we have differentiated the two sets of cross-peaks corresponding to the two members of the dimer with blue and black labels. The cross-peak doubling is particularly obvious for G34 and P25 and also for the important residues H37 and W41. The spectra of Figure 2 illustrate our ability to make consecutive assignments, as shown from P25 to A29. In all, two continuous stretches of backbone assignments were made from these spectra, one for each set of peaks from D24 to D44 and from P25 to Y52. A high magnetic field of 21 T was instrumental in obtaining such wellresolved spectra for assignment.

In Figure 3, we illustrate a 2 D HN spectrum recorded using a ${ }^{1} \mathrm{H}-{ }^{15} \mathrm{~N}$ CP transfer with $\omega_{1} / 2 \pi=50 \mathrm{kHz}$ and $\sim 4-10 \mathrm{kHz}$ on the ${ }^{15} \mathrm{~N}$ and ${ }^{1} \mathrm{H}$ channels, respectively. In addition, we extended this


Figure 1. ${ }^{15} \mathrm{~N}-{ }^{13} \mathrm{C} \alpha$ region of the ZF-TEDOR spectrum of $\mathrm{U}-{ }^{13} \mathrm{C},{ }^{15} \mathrm{~N}$ S31N M2; $\tau_{\text {mix }}=1.2 \mathrm{~ms}$ was used in order to observe only one bond transfer. Note the two sets of cross-peaks (labeled with blue and black) that correspond to the two different molecules of the M2 dimer; $\omega_{0 \mathrm{H}} /$ $2 \pi=900 \mathrm{MHz}$.


Figure 2. Selected strips from the 3D ZF-TEDOR-RFDR spectrum recorded at $\omega_{0 \mathrm{H}} / 2 \pi=900 \mathrm{MHz}$, showing the consecutive assignment of residues P25 to A29. Mixing periods were $\tau_{\text {mix }}=1.2$ and 4.8 ms for ZF-TEDOR and RFDR, respectively, at $\omega_{\mathrm{r}} / 2 \pi=20 \mathrm{kHz}$. The NCACX transfer is shown in red and NCOCX transfer in blue.
experiment to 3 D with an $(\mathrm{H}) \mathrm{C}_{\alpha} \mathrm{NH}$ spectrum ${ }^{37,42}$ recorded using a $\mathrm{H}-\mathrm{C}_{\alpha}$ CP with $\omega_{1} / 2 \pi=35-50 \mathrm{kHz}$ ramp on ${ }^{1} \mathrm{H}$ and 10 kHz on ${ }^{13} \mathrm{C}$, $\mathrm{C}_{\alpha}-\mathrm{N}$ CP with $\omega_{1} / 2 \pi=35 \mathrm{kHz}$ and 25 kHz on $\mathrm{C}_{\alpha}$ and ${ }^{15} \mathrm{~N}$, with a $10 \%$ ramp, and $\mathrm{N}-\mathrm{H}$ CP with $\omega_{1} / 2 \pi=50 \mathrm{kHz}$ on ${ }^{15} \mathrm{~N}$ and $4-10 \mathrm{kHz}$ on ${ }^{1} \mathrm{H}$. Selected strips from the spectrum are shown in Figure 4. Additional 3D $(\mathrm{H}) \mathrm{C}_{\alpha}\left(\mathrm{C}_{0}\right) \mathrm{NH},\left(\mathrm{HC}_{\alpha}\right) \mathrm{C}_{\beta}\left(\mathrm{C}_{\alpha}\right) \mathrm{NH}$, and $\left(\mathrm{HC}_{\alpha}\right)$ $\mathrm{C}_{\beta}\left(\mathrm{C}_{\alpha} \mathrm{C}_{0}\right) \mathrm{NH}$ spectra were also recorded using similar parameters. ${ }^{42}$ The U- ${ }^{13} \mathrm{C},{ }^{15} \mathrm{~N},{ }^{2} \mathrm{H}-\left[{ }^{12} \mathrm{C},{ }^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H} \delta 1\right.$-Ile, ${ }^{12} \mathrm{C},{ }^{13} \mathrm{C}_{\alpha}{ }^{13} \mathrm{C}^{\prime},{ }^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H} \delta 2$ Leu, $\left.{ }^{12} \mathrm{C},{ }^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H} \gamma 2-\mathrm{Val}\right]$ sample was used. The spectra were recorded at $\omega_{0 \mathrm{H}} / 2 \pi=1000 \mathrm{MHz}$ and are an excellent example of a MAS NMR version of a solution NMR HSQC and $\mathrm{HNC}_{\alpha}$ experiment and allow assignment of the ${ }^{15} \mathrm{~N}$ amide resonances as well as structural measurements. Note that the peak doubling is apparent due to the presence of the two conformations in the dimer of dimers.

Measurement of Distance Restraints. Using an ILFY reverselabeled sample, proton-assisted recoupling (PAR) MAS spectra were


Figure 3. Cross-polarization-based HN spectrum showing the completed backbone amide assignments for $\mathrm{M} 2_{18-60}$. Note the cross-peak doubling denoted with blue and black labels on the cross-peaks in the spectrum; $\omega_{\mathrm{r}} / 2 \pi=60 \mathrm{kHz}$ and $\omega_{0 \mathrm{H}} / 2 \pi=1000$ MHz .


Figure 4. Selected strips from the $3 \mathrm{D}(\mathrm{H}) \mathrm{C}_{\alpha} \mathrm{NH}$ spectrum of ${ }^{2} \mathrm{H}$ $\mathrm{M} 2_{18-60}$ back exchanged with ${ }^{1} \mathrm{H}$ at amide sites; $\omega_{0 \mathrm{H}} / 2 \pi=1000 \mathrm{MHz}$.
recorded at $\omega_{0 \mathrm{H}} / 2 \pi=900 \mathrm{MHz}$ and $\omega_{\mathrm{r}} / 2 \pi=20 \mathrm{kHz}$, with $\tau_{\text {mix }}=15$ ms . The spectra, shown in Figure 5, primarily permit identification of $\mathrm{C} \alpha_{i}-\mathrm{C} \alpha_{i+3}$ contacts. In addition, a 4 -fold molar excess of rimantadine was added to the sample and had minimal impact ( $<0.5 \mathrm{ppm}$ ) on the chemical shifts, indicating that the inhibitor does not induce significant conformational changes in the S 31 N mutant. ${ }^{18}$

Proton-driven spin-diffusion (PDSD) spectra $^{43}$ were recorded at $\omega_{0 \mathrm{H}} / 2 \pi=750 \mathrm{MHz}$ and $\omega_{\mathrm{r}} / 2 \pi=14.287 \mathrm{kHz}$, using an ILFY reverselabeled sample with a 4 -fold molar excess of rimantadine. Primarily, $\mathrm{C}_{\alpha}-\mathrm{C}_{\alpha}$ contacts are shown in Figure 6. In addition, cross-peaks arising from structurally important aromatic-aromatic ${ }^{18}$ contacts are present in the spectrum, for example, interhelical contacts from $\mathrm{H} 37 \mathrm{C} \delta 2$ to $\mathrm{H} 37 \mathrm{C} \delta 2^{\prime}$ and from $\mathrm{W} 41 \mathrm{C} \eta 2$ to $\mathrm{W} 41 \mathrm{C} \delta 1^{\prime}$. In these spectra, we do not observe exchange between the two sets of peaks and therefore conclude that the dimer of dimers is stable on the time scale of seconds.


Figure 5. ${ }^{13} \mathrm{C}-{ }^{13} \mathrm{C}$ MAS spectrum of ILFY-labeled $\mathrm{M} 2_{18-60}$ recorded at $\omega_{0 \mathrm{H}} / 2 \pi=900 \mathrm{MHz}$ and $\omega_{\mathrm{r}} / 2 \pi=20 \mathrm{kHz}$ using PAR mixing with $\tau_{\text {mix }}=15 \mathrm{~ms}$. A 4-fold molar excess of rimantadine drug was present in the sample and produced minimal perturbations of the chemical shifts, as expected from an $\mathrm{S} 31 \mathrm{~N} \mathrm{M} 2_{18-60}$.


Figure 6. $\mathrm{C}_{\alpha}-\mathrm{C}_{\alpha}$ region of a PDSD spectrum with $\tau_{\text {mix }}=400 \mathrm{~ms}$ recorded at $\omega_{0 \mathrm{H}} / 2 \pi=750 \mathrm{MHz}$ and $\omega_{\mathrm{r}} / 2 \pi=14.287 \mathrm{kHz}$. A 4-fold molar excess of rimantadine drug was present in the sample and did not perturb the chemical shifts of the S31N mutant.

To establish the structure of the $\mathrm{M} 2_{18-60}$ dimer of dimers, it was important to have interhelical constraints. Accordingly, we prepared a sample with mixed ${ }^{15} \mathrm{~N}$ and ${ }^{13} \mathrm{C}$ labeling and used ZF-TEDOR to detect interhelical contacts. In particular, a $1: 1$ mixture of $\mathrm{U}-{ }^{15} \mathrm{~N}$ $\mathrm{M} 2_{18-60}$ and $1,6-{ }^{13} \mathrm{C}$ glucose-labeled $\mathrm{M} 2_{18-60}$ was refolded, ensuring that mixed tetramers were formed. ZF-TEDOR spectra were recorded at $7 \mu \mathrm{~s}{ }^{15} \mathrm{~N} \pi$ pulses, 100 kHz TPPM ${ }^{44}$ decoupling, at $\omega_{0 \mathrm{H}} / 2 \pi=800$ MHz , using $\tau_{\text {mix }}=7.168,14.336$, and 21.504 ms , with the optimal intensity for most cross-peaks being observed at 14.336 ms . The ZF-

TEDOR experiment permitted identification of five unambiguous and one ambiguous interhelical contact shown in Figure 7.


Figure 7. ${ }^{15} \mathrm{~N}-{ }^{13} \mathrm{C}$ aliphatic correlations from a ${ }^{15} \mathrm{~N}-{ }^{13} \mathrm{C}$ ZF-TEDOR spectrum of a $1: 1$ mixture of ${ }^{15} \mathrm{~N}$ and $1,6-{ }^{13} \mathrm{C}$ glucose-labeled $\mathrm{M} 2_{18-60}$ recorded with $\tau_{\text {mix }}=14.3 \mathrm{~ms}$.

In addition, 10 interhelical contacts (more than any other single experiment) were identified in homonuclear ${ }^{13} \mathrm{C}-{ }^{13} \mathrm{C}$ RFDR spectra of a ${ }^{13} \mathrm{C}$-labeled sample prepared from $1,6-{ }^{13} \mathrm{C}_{2}$ glucose. This produces the same labeling pattern as that in $1-{ }^{13} \mathrm{C}$ glucose ${ }^{33}$ but with twice the site labeling level (approaching $90 \%$ incorporation). Since the ${ }^{13} \mathrm{C}$ are magnetically dilute, the deleterious effects of dipolar truncation are attenuated in measurements of weak couplings. ${ }^{45}$ Two different spectra were recorded with broad-band RFDR using $\tau_{\text {mix }}=8$ and 19.2 ms. In Figure 8, the cross-peaks labeled in red are interhelical and were


Figure 8. ${ }^{13} \mathrm{C}-{ }^{13} \mathrm{C}$ RFDR spectrum of $1,6-{ }^{13} \mathrm{C}$ glucose-labeled $\mathrm{M} 2_{18-60}$ showing interhelical cross-peaks (red labels) inter-residue cross-peaks (green), and intraresidue cross-peaks (black). Gray labels indicate cross-peaks that can only be explained due to the presence of intertetramer contacts. The spectrum was acquired with broad-band RFDR at $\tau_{\text {mix }}=8 \mathrm{~ms}, \omega_{1} / 2 \pi=100 \mathrm{kHz}$ of ${ }^{1} \mathrm{H}$ TPPM decoupling, $\omega_{\mathrm{r}} /$ $2 \pi=20 \mathrm{kHz}$, and $\omega_{0 \mathrm{H}} / 2 \pi=900 \mathrm{MHz}$.
acquired with $\tau_{\text {mix }}=8 \mathrm{~ms}$. Also present in the spectrum are multiple inter- and intraresidue cross-peaks labeled with green and black, respectively. Two additional cross-peaks (gray) are present, which are assigned to intertetramer contacts.

The final two experimental data sets used to constrain distances were obtained from ${ }^{1} \mathrm{H}$-detected 3D and 4D spectra recorded at $\omega_{0 \mathrm{H}} /$ $2 \pi=800 \mathrm{MHz}$ that used ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ RFDR recoupling and provided inter- and intramolecular methyl-methyl and His-Trp distances. The pulse sequence used for acquiring ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ distances between the ${ }^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H}$ groups is similar to a recently reported 4 D
experiment using DREAM mixing ${ }^{46}$ and to a CP-based implementation. ${ }^{47}$ In particular, we used high spinning frequencies $\left(\omega_{\mathrm{r}} / 2 \pi=60\right.$ kHz ) and INEPT for ${ }^{13} \mathrm{C}-{ }^{1} \mathrm{H}$ and ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ transfers. ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ recoupling for the observation of long-range contacts was accomplished using RFDR ( $\omega_{1} / 2 \pi=100 \mathrm{kHz}$ ) with $\tau_{\text {mix }}=8 \mathrm{~ms}$, together with 300 ms of water suppression, as suggested by Rienstra and coworkers. ${ }^{48}$ The timing diagram for the experiment is shown in Figure 9, and selected strips are displayed in Figure 10.


Figure 9. Timing diagram for the $4 \mathrm{D}{ }^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H}$ distance in the HCHHCH spectrum shown in Figure 10. Narrow and wide pulses represent 90 and $180^{\circ}$ pulses, respectively. The delay $\Delta$ was optimized for transfer via the $\mathrm{C}-\mathrm{H} J$-coupling. RFDR and $\mathrm{H}_{2} \mathrm{O}$ suppression blocks were looped to reach the correct time. Low-power TPPM or WALTZ decoupling was used during ${ }^{1} \mathrm{H}$ acquisition. Phases were X unless indicated, and $\phi_{1}=13, \phi_{2}=11113333, \phi_{3}=0022$, and $\phi_{\mathrm{rec}}=$ 02202002.

Finally, a 3D ${ }^{15} \mathrm{~N}-{ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ MAS spectrum using 3.33 ms of ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ RFDR was recorded at $\omega_{\mathrm{r}} / 2 \pi=60 \mathrm{kHz}$ using 100 kHz RFDR pulses. The cross-peak intensities were identified and observed in 2D as the mixing time was varied between 0.267 and 3.33 ms . The spectrum is shown in Figure 13B and highlights the structurally important H37'$\mathrm{H} \varepsilon 2$ to $\mathrm{W} 41-\mathrm{H} \varepsilon 1$ side chain contact in orange. Notably, the intermolecular cross-peak is more intense than similar intramolecular H37-W41 cross-peaks but less intense than $2.8 \AA$ backbone amide contacts. This indicates a close intermolecular approach, consistent with the calculated structure, and is an essential new feature of the dimer of dimers motif.

Conversion of Peak Volumes to Distance Restraints. For ${ }^{13} \mathrm{C}-{ }^{13} \mathrm{C}$ restraints, peak volumes of known distances $\left(\mathrm{C} \alpha_{i}-\mathrm{C} \alpha_{i+1}\right.$ and $\left.\mathrm{C} \alpha_{i}-\mathrm{C} \alpha_{i+3}\right)$ and a distance dependence of $r^{-3}$ were used to calibrate distance restraints. For ${ }^{13} \mathrm{C}-{ }^{13} \mathrm{C}$ RFDR, $\mathrm{C} \alpha_{i}-\mathrm{C} \alpha_{i+1}$ peaks ( $3.8 \AA$ distance) were used for calibration, and an upper distance limit was entered as 1.1 times the calculated distance plus $1 \AA$. For nonaromatic sites, an additional $1 \AA$ was added to account for differences in relaxation. We did not attempt to adjust the distance limits based on the site-specific degree of labeling. For ${ }^{13} \mathrm{C}-{ }^{13} \mathrm{C}$ PDSD and PAR, $\mathrm{C} \alpha_{i}-$ $\mathrm{C} \alpha_{i+3}$ peaks ( $5.3 \AA$ distance) were used for calibration, and an upper distance limit was entered as 1.12 times the calculated distance plus 1 $\AA$. Several restraints from the $\tau_{\text {mix }}=8 \mathrm{~ms}$ PAR spectrum were adjusted manually due to consistent violations. These restraints tended to involve carbonyl resonances, for which relaxation is expected to be different from the proton-bonded carbons used for calibration.

TALOS+ Predictions. TALOS+ predictions were made using all assigned residues both with and without the use of proton chemical shifts. A higher number of good predictions were made without the proton shifts, and we therefore excluded proton shifts in the predictions. This can be rationalized based on the sensitivity of the proton chemical shift to water accessibility, ${ }^{28}$ which is typically low in a membrane protein and perhaps the slightly altered helical geometry ${ }^{49}$ found in membrane proteins that may not be well-represented in the TALOS+ database.

Introduction of Helical Hydrogen Bonds. Based on the TALOS + predictions, helical hydrogen bonds were entered as distance restraints for assigned residues except where the TM and AP helices meet. Distances of $2.0 \pm 0.2 \AA$ for $\mathrm{C}_{i}{ }^{\prime}$ to $\mathrm{HN}_{i+4}$ and $3.0 \pm 0.2 \AA$ for $\mathrm{C}_{i}{ }^{\prime}$ to $\mathrm{N}_{i+4}$ were entered from P 25 to L43. For one pair of helices,


Figure 10. Methyl spectra of $\mathrm{M} 2_{18-60}$ labeled with ${ }^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H}$ groups in the I, L, and V residues. The J-transferred 2D spectrum of the stereospecifically ${ }^{13} \mathrm{C}^{2} \mathrm{H}_{2}{ }^{1} \mathrm{H}$-labeled sample is shown in A , with assignments of isoleucine $\mathrm{C} \delta 1$, leucine $\mathrm{C} \delta 2$, and valine $\mathrm{C} \gamma 2$ methyl groups. Notice the excellent resolution and the peak doubling observed in other ${ }^{13} \mathrm{C} /{ }^{15} \mathrm{~N}$ spectra. $\mathrm{B}-\mathrm{E}$ show selected planes from a ${ }^{1} \mathrm{H}$-detected 4 D using ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ RDFR and $\tau_{\text {mix }}=8 \mathrm{~ms}$ and $J$-couplingbased transfers for ${ }^{13} \mathrm{C}-{ }^{1} \mathrm{H}$ correlation. Interhelical correlations are shown in green. Autocorrelations are indicated with asterisks. Details of the pulse sequence are shown in Figure 9.
resonances were observed for additional residues, and these restraints were therefore continued between residues 47 and 52 .

Highly Ambiguous Restraints. Highly ambiguous restraints were automatically identified in the RFDR spectra of $1,6-{ }^{13} \mathrm{C}$ glucose-labeled $\mathrm{M} 2_{18-60}$ using a cutoff of 0.25 ppm , the assignment tables, and the spectrum. Cross-peaks arising from side bands or truncation artifacts were removed by manual inspection. The restraints were then filtered against a structure generated without these ambiguous restraints but including all manually entered restraints. Restraints that violated by more than $3 \AA$ were removed. The remaining restraints were entered directly into XPLOR as highly ambiguous restraints.

Treatment of Intertetramer Contacts. Contacts that do not fit the general channel architecture (a parallel tetramer with His-37 facing inward) were excluded from the calculation and may be due to intertetramer contacts that arise due to the high concentration of protein in the membrane. For example, Val and Trp are at opposite sides of the membrane, and cross-peaks that are seen between them most likely arise from contacts between adjacent tetramers, which are inserted in an antiparallel arrangement in the membrane. This arrangement is not surprising, given the wedge shape of M2 with a large C-terminal base, and is the arrangement observed in crystal structures of TM M2 ${ }_{22-46}$.

Structure Calculations. Simulated annealing was performed using the program XPLOR-NIH. ${ }^{50}$ Due to the $C_{2}$ symmetry, there are two different interfaces between each helix and its two neighbors. Each restraint was therefore entered as an ambiguous restraint to either of
the neighboring helices. Since use of ambiguous restraints results in a rough energy landscape in which the protein is easily trapped in a local minimum, satisfying one of the possibilities, these ambiguous restraints resulted in slow convergence if introduced too early. First, TALOS+ and hydrogen-bonding restraints were introduced in a standard simulated annealing (SA) protocol, and once the structure converged, the four helices were aligned loosely by applying four-fold symmetric restraints (upper bound $6 \AA$ ) from $\mathrm{D} 44 \mathrm{C} \gamma-\mathrm{R} 45 \mathrm{C} \zeta$, H37C $\varepsilon 1-$ $\mathrm{H} 37 \mathrm{C} \varepsilon 1$, and $\mathrm{V} 27 \mathrm{C} \gamma-\mathrm{V} 27 \mathrm{C} \gamma$. Next, noncrystallographic restraints (NCS) were turned on to ensure $C_{2}$ symmetry, and 33 structures were calculated that satisfied these modeling restraints, TALOS+, and hydrogen-bond restraints. Modeling restraints were turned off and all experimental restraints were included in the final two annealing cycles. First, 30 structures were calculated from each of the 33 structures from the previous annealing. Next, the lowest energy structure from each set of 30 structures was taken as input for calculation of an additional 30 structures. From each of these sets of 30 structures, the lowest energy structure was selected, and the lowest 20 out of 33 final structures were included in the reported ensemble. During annealing, a databasederived potential for side chain rotamers (Rama) was ramped from 0.02 to 0.2 . Flat-well harmonic potentials were used, with force constants ramped from 25 to $100 \mathrm{kcal} \mathrm{mol}^{-1} \AA^{-2}$ for distances (dipole couplings) and 20-100 $\mathrm{kcal} \mathrm{mol}^{-1} \mathrm{rad}^{-2}$ for TALOS-based backbone dihedral restraints. Additional force constants used in the annealing were van der Waals of $0.02-4.0 \mathrm{kcal} \mathrm{mol}{ }^{-1} \AA^{-2}$, improper of $0.1-1.0$ $\mathrm{kcal} \mathrm{mol}{ }^{-1}$ degree $^{-2}$, and bond angle of $0.4-1.0 \mathrm{kcal} \mathrm{mol}^{-1}$ degree $^{-2}$. The temperature was reduced from 1000 to 20 K in steps of 20 K and 4 ps of Verlet dynamics at each temperature with 1.5 fs time steps. The mass of all atoms was set to 100 for the annealing.

## RESULTS

A set of 283 restraints consisting of 70 intrahelical distance measurements, 49 interhelical distance measurements, 72 highly ambiguous distance measurements, and 92 TALOS restraints were used to calculate an atomic resolution structure of $\mathrm{M} 2_{18-60}$ with greater than 5 restraints per residue. Interhelical restraints are depicted in Figure 11 and listed in the Supporting Information. The precision of the structure is characterized by an ensemble of twenty low-energy structures with an rmsd of $0.7 \AA$ for backbone heavy atoms and $1.1 \AA$ for all heavy atoms (Figure 12). The rmsd was calculated for the structured region of the protein, which extended from P25 to R45 and P25 to S50 for the two asymmetric helices, and corresponds to all consistently observed residues. Part of the amphipathic helix was observed for only one set of resonances. The residues that are not observed can be assumed to undergo microsecond to millisecond motion that interferes with the ${ }^{1} \mathrm{H}$ decoupling in MAS experiments and attenuates the intensities. This asymmetry is again consistent with the dimer of dimer structure.

The structure is packed tightly together with a narrow pore. It displays a hydrophobic surface in the direction of lipids (Figure 12) as expected for membrane proteins. At the Cterminal base, the hydrophilic residues of the amphipathic helix are positioned to interact with the hydrophilic head groups of the lipids. In Figure 12C, the interior surface of the tetramer is drawn using the program HOLE. ${ }^{51}$ Consistent with the understanding that the conserved residues H37 and W41 are responsible for ion selectivity and pH dependence, the narrowest part of the channel is found at these residues. The surface is colored red where less than one water molecule can fit in the channel, in green where no more than one water molecule can fit, and in blue where the pore diameter fits multiple water molecules.

Four W41 side chains are found in two distinct secondary structures (Figure 12D), with two helices in an "indole in"


Figure 11. Helical wheel representation of a two-fold symmetric M2 tetramer depicting the set of interhelical restraints (dashed lines) used in the final annealing, after resolution of ambiguities. The side chains of important residues N31, H37, and W41 are indicated with solid black lines.


Figure 12. Dimer of dimers structure of M2. (A) Ensemble of seven low-energy structures. The backbone and all heavy atom rmsd values are 0.7 and $1.1 \AA$, respectively. (B) Surface representation showing a continuous hydrophobic exterior. (C) Pore surface as calculated using the program HOLE, colored in red, green, and blue for pore widths of $<1$ water, 1 water, and $>1$ water, respectively. (D) C-terminal view of the pore with H 37 in red and W 41 in blue in two distinct side chain conformations. W41 adopts an "indole in" conformation for one helix and an "indole out" conformation in the other. Unstructured N - and C-terminal residues are not displayed.
(W41 in Figure 13A) conformation, and the other two in an "indole out" (W41' in Figure 13A) conformation, which in combination with the helix displacement allows the tetramer to pack into a tight bundle with a continues hydrophilic interior. Notably, the M2 channel contains very few hydrophilic residues lining the pore, particularly toward the N -terminus, where a stretch of six hydrophobic residues from P25 to A30 span nearly two helical turns. The result that adjacent helices are out of register decreases the distance between hydrophilic residues, improving the ability of the channel to form a hydrophilic pathway for proton conduction along the pore. The two-fold symmetric structure raises the possibility for an intermolecular mechanism of $\mathrm{H}^{+}$translocation in which the protons first bind to the more N-terminal H37 (Figure 13A) before continuing to the more C-terminal H37. Subsequently, given their proximity in the structure, an intermolecular transfer between H37' and W41 could possibly precede proton exit on the C-terminal side of the protein (Figure 13A and Figure 15). Although this suggestion is consistent with the observed geometry, we do not at present have direct evidence for this pathway.


Figure 13. (A) Assembly of four H 37 and W 41 residues within the dimer of dimers is shown. A short intermolecular distance of $3-3.5 \AA$ is observed between $\mathrm{H} 37^{\prime}$ and W41 in the ensemble of structures calculated from the ${ }^{13} \mathrm{C}-{ }^{15} \mathrm{~N}$ and some of the ${ }^{1} \mathrm{H}$ detected data. (B) This proximity was subsequently confirmed by an (H)NHH NFDR spectrum recorded showing $\mathrm{H}_{3}{ }^{\prime}-\mathrm{W} 41$ cross peaks recorded with $\tau_{\text {mix }}=3.33 \mathrm{~ms}$ of ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ RFDR recoupling. The buildup of this peak is shown for a range of mixing times together with other assigned crosspeaks including a ${ }^{1} \mathrm{H}-{ }^{15} \mathrm{~N}$ backbone distance as a calibration.

## DISCUSSION

Drug Resistance. Despite the S31N mutation occurring near the pharmacological drug-binding site, ${ }^{52,53}$ the precise influence of N31 on drug binding is unclear in the literature due to the substantial differences in reported structures that were solved for a variety of constructs under sample conditions that may not adequately mimic the viral membrane. There are
two reported structures of the S31N mutant. The first of these was determined under conditions that did not bind drug in the pore of WT. ${ }^{20}$ The other was solved with an inhibitor different from amantadine and rimantadine. ${ }^{21}$ While previous structures have pointed toward either helix packing ${ }^{9,20}$ or direct interaction ${ }^{10}$ to explain drug resistance, the two-fold symmetric structure contains elements of both. In particular, the side chain of N31 points toward the pore in two helices and toward an adjacent helix in the other two, with neighboring N31 side chains close enough to form polar contacts. N31 is wellshielded from contact with the hydrophobic lipid membrane. In contrast, drug-bound structures of $\mathrm{M}^{9,26}$ show position 31 in the interface between two helices, with the side chain oriented toward the lipids. This implies that two of the helices of S31N must rotate for the molecule to adopt the drug-bound structure (see Figure 14). In S31N, the larger hydrophilic N31 side chain


Figure 14. S31N and drug-bound structures explain drug resistance.The S31N structure in lipids (A) places N31 in the pore or in the helix-helix interface. The WT chimera structure (B) with bound drug (pdb code 2LJC) places S31 in the helix-helix interface pointing out toward the lipid membrane mimetic. Drug is shown in magenta, the drug-binding pocket (V27 and A30) is depicted as green sticks, and residue 31 (serine or asparagine) is shown as spheres.
would disfavor this motion due to interaction with hydrophobic lipids. In addition, a direct interaction between the N31 side chain in the pore and the drug would be unfavorable. While both effects could contribute to resistance, it is unclear which is more important. The simplest explanation is that in the drugbound state position 31 points toward lipids which is favorable for serine but not asparagine. Thus, the new dimer of dimer structure of S31N shows the necessary rearrangement of the N31 orienting it away from the lipids. According to the structure of the rimantadine-bound chimera M2, this S31N structure clearly does not contain the adamantane binding pocket. For the design of a novel inhibitor, it may be useful to block the channel in the apo structure, without a helix rotation. The structure presented herein should find use in the search for novel inhibitors, which has recently identified several promising compounds. ${ }^{25,29,30,54}$ As evident in Figure 12, the largest pocket in the pore that might be targeted by an inhibitor is the pharmacological binding site of adamantane-based drugs, near residue 31.

Regulation of the pH -dependent proton conduction and selectivity is determined largely by the conserved HxxxW motif, with the rate of conduction tuned by the $\mathrm{p} K_{\mathrm{a}}$ of $\mathrm{H} 37^{14,15}$ and the unidirectional flow of protons controlled by W41. ${ }^{55}$ Previous explanations of the unusual $\mathrm{p} K_{\mathrm{a}}$ of H 37 led to the proposal of imidazole-imidazolium dimerization in which the
first two protons to enter the tetramer were shared in a low barrier hydrogen bond (LBHB). ${ }^{15}$ An alternative explanation places H37 in a His-box conformation, ${ }^{27}$ with water as the hydrogen-bonding partner. ${ }^{14}$ The present structure deviates from a His-box conformation, but since it was solved at pH 7.8 , above the first $\mathrm{p} K_{\mathrm{a}}$ of this construct, and therefore does not rule out a LBHB at lower pH . If the dimerization does occur, some rearrangement of the helices would be expected in order to bring the $\delta 1$ and $\varepsilon 2$ nitrogens of adjacent H37s near enough to form a hydrogen bond ( $\sim 5 \AA$ in the structure). Alternatively, tuning of the $\mathrm{H} 37 \mathrm{p} K_{\mathrm{a}}$ could be accomplished with favorable cation $-\pi$ interactions. The two-fold symmetric structure has a network of $\pi$ systems that could form favorable cation $-\pi$ interactions between histadine and tryptophan, both within a helix, and between neighboring helices.

As noted above, the two-fold symmetric structure of the tetramer reveals that the side chain $\mathrm{N}-\mathrm{H}$ groups of $\mathrm{H} 37^{\prime}$ and W41 are in close proximity (Figure 13A). This feature emerged in the initial structure calculations and suggested the possibility that it could be confirmed with ${ }^{1} \mathrm{H}$ detected MAS experiments. Thus, we performed a $(\mathrm{H}) \mathrm{NHH}_{\text {RFDR }}$ experiment which yielded the $2 \mathrm{D}{ }^{1} \mathrm{H}-{ }^{15} \mathrm{~N}$ spectrum illustrated in Figure 13B which shows strong $\mathrm{H} 37^{\prime}-\mathrm{W} 41$ cross peaks providing an $\mathrm{N}-\mathrm{N}$ distance constraint of $<3.5 \AA$. The proximity of these two residues suggests that an intermolecular proton transfer might be involved in the conduction, and a possible mechanism entailing this process is depicted in Figure 15. Initially, H37' is


Figure 15. A possible conduction mechanism consistent with the close proximity of $\mathrm{H} 37^{\prime}$ and W41 found in the dimer of dimers structure (Figure 13A). The black H represents a proton that enters the channel from the n -terminal side. The blue H represents a proton transferred from H 47 to W41. The red H represents the proton that can be readily released into the virus. Tautomerization of the H37 side-chain brings the protein back to the original state.
protonated, and the transfer from H37' through the channel is mediated via W41. An interesting question is then what is the function of the other two H37's and W41's in this mechanism. If significant structural changes occurred along with the proton release via W 41 then, the other pair of intermonomer H37-W41 contacts might form. So it is possible that a proton relay might involve alternating formation of intermonomer H37-W41 complexes. However, the second pair of H37-W41's are some distance removed and this would then average the two sets of cross peaks observed in the spectra. Clearly additional experiments are required to resolve these interesting questions.

Other Structures of M2. M2 structural models and structures have been determined under a variety of sample conditions, including both lipid bilayers, in detergent micelles, and crystallized from detergents. Among the complete tetramer structures, these investigations have either been conducted in
detergents or have included a limited set of interhelical restraints, resulting in reported structures that show a high degree of conformational variability, which must be explained primarily from the difference in sample preparations. Surprisingly, none of the tetramer structures reported thus far are a dimer of dimers, as was recently observed in lipids via peak doubling, ${ }^{18-20}$ except for a recent structure ${ }^{56}$ based on oriented sample NMR and molecular dynamics simulations in which the deviation from four-fold symmetry was restricted to the side chain of H37.

In this oriented sample NMR structure, the helical tilt angle was determined experimentally, and a tetramer was assembled using molecular dynamics simulations. However, such orientation constraints ${ }^{57}$ are invariant with respect to translation and rotation about the bilayer normal. Since the four-fold symmetry of the backbone is primarily broken by translation in the S31N structure in lipids, this structure is in general agreement with the previously reported oriented sample measurements of WT M2, although there may be some differences due to the mutation. This highlights the importance of measuring interhelical contacts in the determination of structure. Ideally, restraints from oriented samples would also be included in structure determinations because orientation restraints can provide long-range information that is complementary to distance measurements. However, currently, such data are not available for the S31N mutant, and we instead used an overdetermined set of distance restraints. Finally, we should mention that the possibility of a dimeric structures for various constructs of M2 has emerged from two recent MD simulations. ${ }^{58,59}$

## ■ CONCLUSION

We have determined the structure of the drug-resistant S31N mutant of M2 in lipid bilayers. The experimental protocol, particularly the experiments to detect interhelical distances, could represent a paradigm for structural determinations of membrane protein samples. The side chain of the mutated residue occludes the binding site of inhibitors in the $n$-terminal pore, explaining drug resistance. The protein was found to adopt a dimer of dimers structure, with significant deviation from two-fold symmetry, particularly for functionally important residues H37 and W41. A short distance of $<3.5 \AA$ was found between polar side chain ${ }^{15} \mathrm{~N}$ of these H and W residues, suggesting that proton conduction may progress by direct transfer from histidine to tryptophan.

## - ASSOCIATED CONTENT

## (s) Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/jacs.5b04802.

Complete description of the synthesis of deuterated DPhPC lipid, a table of assignments, and a complete structural restraint list (PDF)

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## Notes

The authors declare no competing financial interest.

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## REFERENCES

(1) Holsinger, L. J.; Alams, R. Virology 1991, 183, 32.
(2) Pinto, L. H.; Holsinger, L. J.; Lamb, R. A. Cell 1992, 69, 517.
(3) Maeda, T.; Ohnishi, S. FEBS Lett. 1980, 122, 283.
(4) White, J.; Matlin, K.; Helenius, A. J. Cell Biol. 1981, 89, 674.
(5) Zebedee, S. L.; Richardson, C. D.; Lamb, R. A. J. Virol 1985, 56, 502.
(6) Lamb, R. A.; Zebedee, S. L.; Richardson, C. D. Cell 1985, 40, 627.
(7) Tobler, K.; Kelly, M. L.; Pinto, L. H.; Lamb, R. A. J. Virol 1999, 73, 9695.
(8) Andreas, L. B.; Barnes, A. B.; Corzilius, B.; Chou, J. J.; Miller, E. A.; Caporini, M.; Rosay, M.; Griffin, R. G. Biochemistry 2013, 52, 2774.
(9) Pielak, R. M.; Oxenoid, K.; Chou, J. J. Structure 2011, 19, 1655.
(10) Stouffer, A. L.; Acharya, R.; Salom, D.; Levine, A. S.; Di Costanzo, L.; Soto, C. S.; Tereshko, V.; Nanda, V.; Stayrook, S.; DeGrado, W. F. Nature 2008, 451, 596.
(11) Cady, S. D.; Schmidt-Rohr, K.; Wang, J.; Soto, C. S.; DeGrado, W. F.; Hong, M. Nature 2010, 463, 689.
(12) Cady, S. D.; Wang, J.; Wu, Y.; DeGrado, W. F.; Hong, M. J. Am. Chem. Soc. 2011, 133, 4274.
(13) Schnell, J.; Chou, J. Nature 2008, 451, 591.
(14) Hu, F.; Schmidt-Rohr, K.; Hong, M. J. Am. Chem. Soc. 2012, 134, 3703.
(15) Hu, J.; Fu, R.; Nishimura, K.; Zhang, L.; Zhou, H. X.; Busath, D. D.; Vijayvergiya, V.; Cross, T. A. Proc. Natl. Acad. Sci. U. S. A. 2006, 103, 6865.
(16) Colvin, M. T.; Andreas, L. B.; Chou, J. J.; Griffin, R. G. Biochemistry 2014, 53, 5987.
(17) Munowitz, M.; Bachovchin, W. W.; Herzfeld, J.; Dobson, C. M.; Griffin, R. G. J. Am. Chem. Soc. 1982, 104, 1192.
(18) Andreas, L. B.; Eddy, M. T.; Chou, J. J.; Griffin, R. G. J. Am. Chem. Soc. 2012, 134, 7215.
(19) Andreas, L. B.; Eddy, M. T.; Pielak, R. M.; Chou, J.; Griffin, R. G. J. Am. Chem. Soc. 2010, 132, 10958.
(20) Can, T. V.; Sharma, M.; Hung, I.; Gor'kov, P. L.; Brey, W. W.; Cross, T. A. J. Am. Chem. Soc. 2012, 134, 9022.
(21) Kovacs, F. A.; Cross, T. A. Biophys. J. 1997, 73, 2511.
(22) Kovacs, F. A.; Denny, J. K.; Song, Z.; Quine, J. R.; Cross, T. A. J. Mol. Biol. 2000, 295, 117.
(23) Song, Z. Y.; Kovacs, F. A.; Wang, J.; Denny, J. K.; Shekar, S. C.; Quine, J. R.; Cross, T. A. Biophys. J. 2000, 79, 767.
(24) Zhou, H.-X.; Cross, T. A. Annu. Rev. Biophys. 2013, 42, 361.
(25) Williams, J. K.; Tietze, D.; Wang, J.; Wu, Y.; DeGrado, W. F.; Hong, M. J. Am. Chem. Soc. 2013, 135, 9885.
(26) Pielak, R. M.; Schnell, J. R.; Chou, J. J. Proc. Natl. Acad. Sci. U. S. A. 2009, 106, 7379.
(27) Acharya, R.; Carnevale, V.; Fiorin, G.; Levine, B. G.; Polishchuk, A. L.; Balannik, V.; Samish, I.; Lamb, R. A.; Pinto, L. H.; DeGrado, W. F.; Klein, M. L. Proc. Natl. Acad. Sci. U. S. A. 2010, 107, 15075.
(28) Walsh, S. T. R.; Cheng, R. P.; Wright, W. W.; Alonso, D. O. V.; Daggett, V.; Vanderkooi, J. M.; DeGrado, W. F. Protein Sci. 2003, 12, 520.
(29) Wang, J.; Wu, Y.; Ma, C.; Fiorin, G.; Pinto, L. H.; Lamb, R. A.; Klein, M. L.; Degrado, W. F. Proc. Natl. Acad. Sci. U. S. A. 2013, 110, 1315.
(30) Wang, J. Z.; Ma, C. L.; Wang, J.; Jo, H.; Canturk, B.; Fiorin, G.; Pinto, L. H.; Lamb, R. A.; Klein, M. L.; DeGrado, W. F. J. Med. Chem. 2013, 56, 2804.
(31) Vuister, G. W.; Kim, S. J.; Wu, C.; Bax, A. J. Am. Chem. Soc. 1994, 116, 9206.
(32) Atreya, H. S.; Chary, K. V. R. Curr. Sci. India 2000, 79, 504.
(33) Lundstrom, P.; Teilum, K.; Carstensen, T.; Bezsonova, I.; Wiesner, S.; Hansen, D. F.; Religa, T. L.; Akke, M.; Kay, L. E. J. Biomol. NMR 2007, 38, 199.
(34) Rosen, M. K.; Gardner, K. H.; Willis, R. C.; Parris, W. E.; Pawson, T.; Kay, L. E. J. Mol. Biol. 1996, 263, 627.
(35) Gans, P.; Hamelin, O.; Sounier, R.; Ayala, I.; Dura, M. A.; Amero, C. D.; Noirclerc-Savoye, M.; Franzetti, B.; Plevin, M. J.; Boisbouvier, J. Angew. Chem., Int. Ed. 2010, 49, 1958.
(36) Volk, J.; Herrmann, T.; Wuthrich, K. J. Biomol. NMR 2008, 41, 127.
(37) Barbet-Massin, E.; Pell, A. J.; Retel, J. S.; Andreas, L. B.; Jaudzems, K.; Franks, W. T.; Nieuwkoop, A. J.; Hiller, M.; Higman, V.; Guerry, P.; Bertarello, A.; Knight, M. J.; Felletti, M.; Le Marchand, T.; Kotelovica, S.; Akopjana, I.; Tars, K.; Stoppini, M.; Bellotti, V.; Bolognesi, M.; Ricagno, S.; Chou, J. J.; Griffin, R. G.; Oschkinat, H.; Lesage, A.; Emsley, L.; Herrmann, T.; Pintacuda, G. J. Am. Chem. Soc. 2014, 136, 12489.
(38) Jaroniec, C. P.; Filip, C.; Griffin, R. G. J. Am. Chem. Soc. 2002, 124, 10728.
(39) Daviso, E.; Eddy, M. T.; Andreas, L. B.; Griffin, R. G.; Herzfeld, J. J. Biomol. NMR 2013, 55, 257.
(40) Bennett, A. E.; Rienstra, C. M.; Griffiths, J. M.; Zhen, W. G.; Lansbury, P. T.; Griffin, R. G. J. Chem. Phys. 1998, 108, 9463.
(41) Bennett, A. E.; Ok, J. H.; Griffin, R. G.; Vega, S. J. Chem. Phys. 1992, 96, 8624.
(42) Knight, M. J.; Webber, A. L.; Pell, A. J.; Guerry, P.; BarbetMassin, E.; Bertini, I.; Felli, I. C.; Gonnelli, L.; Pierattelli, R.; Emsley, L.; Lesage, A.; Herrmann, T.; Pintacuda, G. Angew. Chem., Int. Ed. 2011, 50, 11697.
(43) Szeverenyi, N. M.; Sullivan, M. J.; Maciel, G. E. J. Magn. Reson. 1982, 47, 462.
(44) Bennett, A. E.; Rienstra, C. M.; Auger, M.; Lakshmi, K. V.; Griffin, R. G. J. Chem. Phys. 1995, 103, 6951.
(45) Hohwy, M.; Rienstra, C. M.; Jaroniec, C. P.; Griffin, R. G. J. Chem. Phys. 1999, 110, 7983.
(46) Huber, M.; Hiller, S.; Schanda, P.; Ernst, M.; Bockmann, A.; Verel, R.; Meier, B. H. ChemPhysChem 2011, 12, 915.
(47) Linser, R.; Bardiaux, B.; Andreas, L. B.; Hyberts, S. G.; Morris, V. K.; Pintacuda, G.; Sunde, M.; Kwan, A. H.; Wagner, G. J. Am. Chem. Soc. 2014, 136, 11002.
(48) Zhou, D. H.; Rienstra, C. M. J. Magn. Reson. 2008, 192, 167.
(49) Olivella, M.; Deupi, X.; Govaerts, C.; Pardo, L. Biophys. J. 2002, 82, 3207.
(50) Schwieters, C. D.; Kuszewski, J. J.; Tjandra, N.; Clore, G. M. J. Magn. Reson. 2003, 160, 65.
(51) Smart, O. S.; Goodfellow, J. M.; Wallace, B. A. Biophys. J. 1993, 65, 2455.
(52) Hay, A. J.; Zambon, M. C.; Wolstenholme, A. J.; Skehel, J. J.; Smith, M. H. J. Antimicrob. Chemother. 1986, 18, 19.
(53) Bright, R. A.; Shay, D. K.; Shu, B.; Cox, N. J.; Klimov, A. I. JAMA 2006, 295, 891.
(54) Kolocouris, A.; Tzitzoglaki, C.; Johnson, F. B.; Zell, R.; Wright, A. K.; Cross, T. A.; Tietjen, I.; Fedida, D.; Busath, D. D. J. Med. Chem. 2014, 57, 4629.
(55) Tang, Y.; Zaitseva, F.; Lamb, R. A.; Pinto, L. H. J. Biol. Chem. 2002, 277, 39880.
(56) Sharma, M.; Yi, M.; Dong, H.; Qin, H.; Peterson, E.; Busath, D. D.; Zhou, H. X.; Cross, T. A. Science 2010, 330, 509.
(57) Wu, C. H.; Ramamoorthy, A.; Opella, S. J. J. Magn. Reson., Ser. A 1994, 109, 270.
(58) Carpenter, T.; Bond, P. J.; Khalid, S.; Sansom, M. S. P. Biophys. J. 2008, 95, 3790.
(59) Chen, H.; Wu, Y.; Voth, G. A. Biophys. J. 2007, 93, 3470.


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